

**Figure: 30 TAC §112.233(d)(3)(A)**

$$SO_2 = F_{sc} \times FFa \times \frac{T_{sc}}{T_a} \times \frac{P_a}{P_{sc}} \times \frac{lb\ mole}{385.27\ scf} \times \frac{64.06\ lb\ SO_2}{lb\ mole}$$

Where:

$SO_2$  = affected combustion equipment sulfur dioxide emissions in pounds per hour;

$F_{sc}$  = fuel total sulfur concentration in cubic feet per 1,000,000 cubic feet of flared gas;

$FFa$  = fuel flow in actual cubic feet per hour;

$P_{sc}$  = regulatory standard condition pressure of 14.7 pounds per square inch (psia);

$P_a$  =  $FFa$  measurement pressure in units of psia;

$T_{sc}$  = regulatory standard condition temperature of 528 degrees Rankin; and

$T_a$  = fuel temperature in degrees Rankin.